



# Raman scattering from ZnO incorporating Fe nanoparticles: Vibrational modes and low-frequency acoustic modes

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## ABSTRACT

Nanocrystalline samples of ZnO(Fe) were synthesized by wet chemical method. Samples were characterized by X-ray diffraction to determine composition of the samples (ZnO, Fe<sub>2</sub>O<sub>3</sub>, ZnFe<sub>2</sub>O<sub>4</sub>) and the mean crystalline size (8–52 nm). In this paper we report the experimental spectra of Raman scattering. Main characteristics of experimental Raman spectrum in 200–1600 cm<sup>-1</sup> spectral region are: sharp peak at 436 cm<sup>-1</sup> and broad two-phonon structure at ~1150 cm<sup>-1</sup>, typical for ZnO; broad structure below 700 cm<sup>-1</sup> that has different position and shape in case of ZnFe<sub>2</sub>O<sub>4</sub> or Fe<sub>2</sub>O<sub>3</sub> nanoparticles. Low-frequency Raman modes were measured and assigned according to confined acoustic vibrations of spherical nanoparticles. Frequencies of these vibrational modes were analyzed in elastic continuum approximation, which considers nanoparticle as homogeneous elastic sphere.

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## 1. Introduction

Recently, a continued interest in the synthesis and properties of nanoscale inorganic materials has been observed. Semiconductor nanocrystals have been of much interest over two decades because of their unique physical properties resulting from modification of the electronic states due to the confinement effect. Currently, nanostructures made of ZnO have attracted significant attention owing to their proposed applications in low-voltage and short-wavelength electro-optical devices, transparent ultraviolet protection films, and spintronic devices [1,2].

A considerable attention has recently been devoted to high temperature ferromagnetism observed in transition metal doped oxides. Particularly ZnO has been identified as a promising host semiconductor material, exhibiting ferromagnetism when doped with most of the transition metals – V, Cr, Fe, Co, Ni [3]. However, the origin of ferromagnetic behavior is not very well known in these compounds. Recently, it was shown that the ferromagnetism in these materials can be induced by inclusions of nanoscale oxides of transition metals [4] and/or nanoparticles containing a large concentration of magnetic ions [5]. Novel methods enabling a control of nanoassembling of magnetic nanocrystals in noncon-

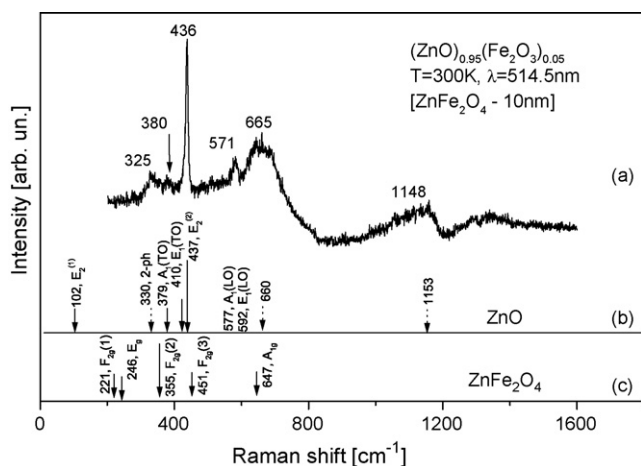
ducting matrices as well as functionalities specific to such systems were described [5].

## 2. Samples and characterization

The nanocrystalline samples of ZnO doped Fe<sub>2</sub>O<sub>3</sub> were synthesized by co-precipitation method [6]. First, the mixture of iron hydroxides and zinc hydroxides from an aqueous solution of nitrites were obtained. Next, the obtained hydroxides were filtered, dried at the temperature of 70 °C and calcined at 300 °C during 1 h. The series of samples containing from 5 to 95 wt.% of Fe<sub>2</sub>O<sub>3</sub> was obtained. In the present paper we present the results of micro-Raman studies and low-frequency Raman modes measurements for two selected samples with low (5 wt.%) and high (90 wt.%) content of magnetic dopant.

The structural properties of the samples were studied by means X-ray diffraction technique (CoK<sub>α</sub> radiation, X'Pert Philips). The performed XRD measurements allowed for the phase composition determination and to determine a mean crystallite size *d* in prepared samples by use of Scherrer's formula [7]. The detailed studies of structural properties by means of XRD technique were presented in Ref. [6]. The crystalline phases of hexagonal ZnO, rhombohedral Fe<sub>2</sub>O<sub>3</sub> and cubic ZnFe<sub>2</sub>O<sub>4</sub> were identified as presented in [6]. The mean crystalline size of these phases is given in Table 1 in Ref. [6]. For sample with 5 wt.% of Fe<sub>2</sub>O<sub>3</sub>, assigned as (ZnO)<sub>0.95</sub>(Fe<sub>2</sub>O<sub>3</sub>)<sub>0.05</sub>, crystalline phase of ZnFe<sub>2</sub>O<sub>4</sub> is identified. Mean crystallite size is

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**Fig. 1.** Raman spectra of nanocrystalline ZnO doped with 5 wt.% of  $\text{Fe}_2\text{O}_3$  assigned as  $(\text{ZnO})_{0.95}(\text{Fe}_2\text{O}_3)_{0.05}$ .

$a = 10$  nm. The second sample with 90 wt.% of  $\text{Fe}_2\text{O}_3$  is assigned as DS90Fe10Zn, crystallite of  $\text{Fe}_2\text{O}_3$  is identified with mean crystallite size  $a = 24$  nm. No other crystal phases are observed in these samples.

XRD measurements did not reveal presence of ZnO phase in these samples. It is interesting that for the samples with lower concentration of  $\text{Fe}_2\text{O}_3$  (up to 20 wt.% of  $\text{Fe}_2\text{O}_3$ ) excitonic lines from ZnO are observed. In some samples, for instance sample containing 40 wt.% of  $\text{Fe}_2\text{O}_3$ , crystallites of ZnO and  $\text{ZnFe}_2\text{O}_4$  are identified [6].

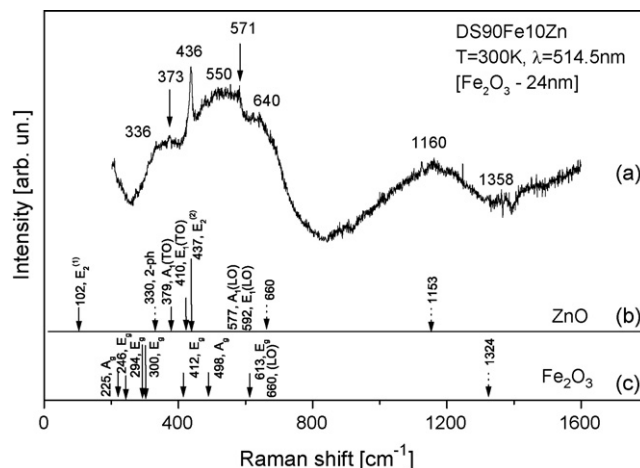
### 3. Results and discussion

#### 3.1. Raman spectroscopy

The micro-Raman spectra were taken in the backscattering configuration and analyzed using Jobin Yvon T64000 spectrometer, equipped with nitrogen cooled charge-coupled-device detector. As excitation source we used the 514.5 nm line of an Ar–iron laser. The measurements were performed at different laser power. Fig. 1(a) Raman spectra of nanocrystalline ZnO doped with 5 wt.% of  $\text{Fe}_2\text{O}_3$  assigned as  $(\text{ZnO})_{0.95}(\text{Fe}_2\text{O}_3)_{0.05}$  is presented.

We will start analysis of spectrum in Fig. 1 with brief report about structure and vibrational properties of potentially present materials in sample. Vibrational properties of bulk material are crucial for understanding vibrational properties of small particles. As consequence of miniaturization, we expect bulk modes to be shifted and broadened.

Basic material in this research is ZnO. ZnO is a semiconductor with a wurtzite crystal structure. This hexagonal structure belongs to the space group  $C_{6v}^4$ , with two formula units per primitive cell, where all atoms occupy  $C_{3v}$  sites. The optical phonons at the center of the Brillouin zone belong to the following irreducible representations:  $\Gamma_{\text{opt}} = A_1 + 2B_1 + E_1 + 2E_2$ . Modes of symmetry  $A_1$ ,  $E_1$  and  $E_2$  are Raman active, and  $A_1$  and  $E_1$  are infrared active.  $B_1$  are inactive (silent) modes. Both  $A_1$  and  $E_1$  are polar modes and split into transverse (TO) and longitudinal (LO) phonons with different frequencies due to macroscopic electric fields associated with the LO phonons. The short-range interatomic forces cause anisotropy. That is why  $A_1$  and  $E_1$  modes have different frequencies. As the electrostatic forces dominate the anisotropy in the short-range forces, the TO–LO splitting is larger than the  $A_1$ – $E_1$  splitting. For the lattice vibration  $A_1$  atoms move parallel to the  $c$ -axis and for  $E_1$  perpendicular to  $c$ -axis. Two nonpolar Raman active modes, are often assigned as  $E_2^{(1)}$ (low) and  $E_2^{(2)}$ (high). All these modes have been reported in the Raman scattering spectra of bulk ZnO many times



**Fig. 2.** Raman spectra of nanocrystalline ZnO doped with 90 wt.% of  $\text{Fe}_2\text{O}_3$  assigned as DS90Fe10Zn.

[8,9]. Frequencies and assignment of Raman active modes in ZnO are presented on the bottom of Fig. 1(b). Solid lines indicate  $A_1$ ,  $E_1$  and  $E_2$  phonon modes in ZnO. Dashed lines mark multi-phonon scattering at  $\sim 330$   $\text{cm}^{-1}$ ,  $\sim 655$   $\text{cm}^{-1}$  and  $\sim 1153$   $\text{cm}^{-1}$  (2LO).

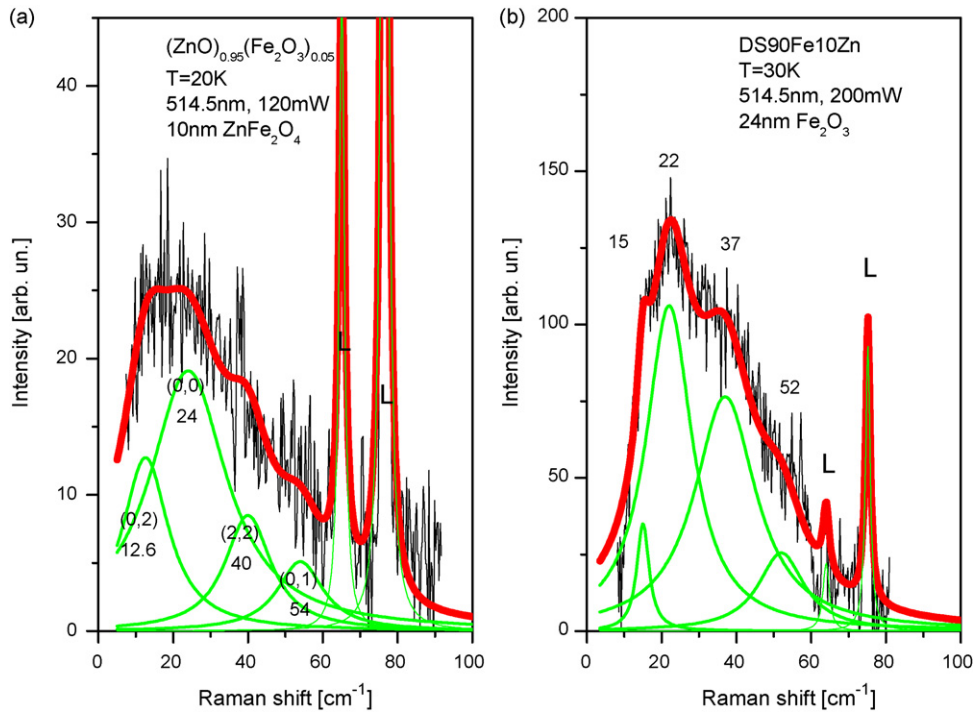
$\text{ZnFe}_2\text{O}_4$  spinel has a cubic structure that belongs to the space group  $O_h^7$ . Full unit cell contains eight formula units, and primitive cell contains two formula units. The optical phonons at  $\Gamma$ -point of the Brillouin zone belong to the following irreducible representations:  $\Gamma_{\text{opt}} = A_{1g} + E_g + F_{1g} + 3F_{2g} + 2A_u + 2E_u + 4F_{1u} + 2F_{2u}$ . There are five first order active Raman modes:  $A_{1g}$ ,  $E_g$  and  $3F_{2g}$ . Modes  $F_{1u}$  are infrared-active. In the cubic spinels, including ferrites, the modes above 600  $\text{cm}^{-1}$  mostly correspond to the motion of oxygen atoms in tetrahedral  $\text{AO}_4$  group. The other lower frequency modes represent the characteristic of the octahedral  $\text{BO}_6$  sites. Frequencies [10] and assignments of Raman-active modes in  $\text{ZnFe}_2\text{O}_4$  are presented on the bottom of Fig. 1(c).

Dispute the fact that XRD does not evident ZnO, sharp peak in our spectrum at 436  $\text{cm}^{-1}$  is clearly  $E_2^{(2)}$  mode of ZnO. Position is practically the same as in bulk material. Phonon dispersion relation, that corresponds to the  $E_2^{(2)}$  mode in ZnO at the  $\Gamma$  point, does not show considerable dispersion. So, even in ZnO nanoparticle spectra position of this mode is practically unchanged. Peak at  $\sim 571$   $\text{cm}^{-1}$ , is identified as LO ZnO mode. Also structure at  $\sim 1150$   $\text{cm}^{-1}$  can be prescribed to multiphonon 2LO mode in ZnO. Wide and dominant structure centered at 665  $\text{cm}^{-1}$  corresponds to vibrations in the tetrahedral  $\text{AO}_4$  group. Materials like:  $\text{Mn}_3\text{O}_4$ ,  $\text{Fe}_3\text{O}_4$ ,  $\text{Co}_3\text{O}_4$  have Raman active mode in interval 600–700  $\text{cm}^{-1}$ . Even in bulk  $\text{ZnFe}_2\text{O}_4$  three the most intensive Raman modes at 355  $\text{cm}^{-1}$ , 451  $\text{cm}^{-1}$  and 647  $\text{cm}^{-1}$  exhibit broad characteristics. If there is any disorder of the Zn and Fe cations in the tetrahedral and octahedral sites, the vibrations related to the two types of cations at the same site may result in two separated first order Raman modes. If they are very close in frequency an overlapped broad peak should be observed.

Raman spectrum in region below 400  $\text{cm}^{-1}$  is superposition of characteristic frequencies of ZnO phase and spectrum of  $\text{ZnFe}_2\text{O}_4$  nanoparticles. In bulk  $\text{ZnFe}_2\text{O}_4$  mode at 355  $\text{cm}^{-1}$  is strong and wide, and modes at 221  $\text{cm}^{-1}$  and 246  $\text{cm}^{-1}$  are weak. In  $\text{ZnFe}_2\text{O}_4$  nanoparticles these modes are wider and contribute to high level of spectral curve.

Raman spectra of nanocrystalline ZnO doped with 90 wt.% of  $\text{Fe}_2\text{O}_3$  assigned as DS90Fe10Zn is presented in Fig. 2. Only nanoparticles of  $\text{Fe}_2\text{O}_3$  are registered by XRD.

We will give brief report about structure and vibrational properties of  $\text{Fe}_2\text{O}_3$ .  $\text{Fe}_2\text{O}_3$  crystallize in the rhombohedral (trigonal) system with space group  $D_{3d}^6$ . Primitive unit cell contains two



**Fig. 3.** Low-frequency Raman spectra of sample: (a)  $(\text{ZnO})_{0.95}(\text{Fe}_2\text{O}_3)_{0.05}$  where  $\text{ZnFe}_2\text{O}_4$  crystallites of  $a = 10$  nm size are identified, (b) DS90Fe10Zn where  $\text{Fe}_2\text{O}_3$  crystallites of  $a = 24$  nm size are identified.

formula units. The optical phonons at the  $\Gamma$ -point of the Brillouene zone belong to the following irreducible representations:  $\Gamma = 2A_{1g} + 2A_{1u} + 3A_{2g} + 2A_{2u} + 5E_g + 4E_u$ .  $2A_{1g}$  and  $5E_g$  are Raman active modes.  $2A_{2u}$  and  $4E_u$  modes are infrared active. Information about  $\text{Fe}_2\text{O}_3$  Raman spectra are presented at the bottom of Fig. 2(c) [11–13]. In addition to these first order Raman spectra there is a very prominent peak at  $\sim 1320 \text{ cm}^{-1}$ . LO phonon situated at  $\sim 660 \text{ cm}^{-1}$  is Raman forbidden. It can be activated, and become visible, if symmetry rules are broken. The 2LO mode which is Raman allowed is seen in all the spectra. Strength of this mode is remarkable in case of resonance.

ZnO mode at  $436 \text{ cm}^{-1}$ , LO ZnO mode at  $\sim 571 \text{ cm}^{-1}$  and ZnO multiphonon mode at  $\sim 1160 \text{ cm}^{-1}$  can easily be identified. In spectral region below  $400 \text{ cm}^{-1}$  there is superposition of ZnO phase and  $\text{Fe}_2\text{O}_3$  nanoparticles spectrum. Difference of spectra in Figs. 1 and 2, in region below  $400 \text{ cm}^{-1}$ , is consequence of presence of  $\text{ZnFe}_2\text{O}_4$  (Fig. 1) or  $\text{Fe}_2\text{O}_3$  (Fig. 2) nanoparticles.

### 3.2. Low-frequency Raman spectroscopy

The Raman spectra were excited by the 514.5 nm line of an Ar laser (the average power was about 120 mW) in the backscattering geometry. We used Jobin Yvon model U-1000 monochromator, with a conventional photocounting system. In order to make low-frequency Raman modes visible we subtracted the intensity of the elastic scattering background by  $A/\omega^n$  approximation. Resulted Raman spectra are presented in Fig. 3. The observed modes are deconvoluted using the Lorentzian line profile.

Low-frequency modes are analyzed as confined acoustic vibrations of nanoparticles. The frequencies of acoustic vibrational modes can be calculated in the elastic continuum approximation. Parameters of the model are stiffness constants and mass density of the particle material. In doped samples  $\text{ZnFe}_2\text{O}_4$  and  $\text{Fe}_2\text{O}_3$  structures are identified. As the dimension of the particles is already determined and having in mind specific frequency to diameter dependence, we established material parameters and analyzed behavior of nanoparticles.

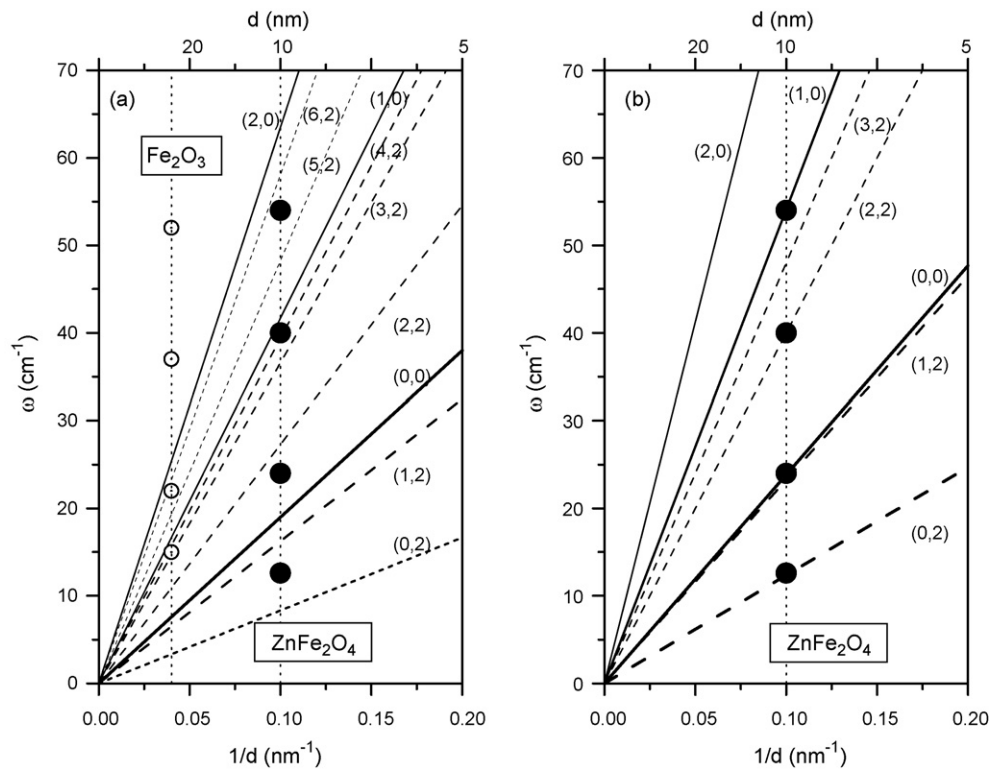
Transverse ( $v_T$ ) and longitudinal ( $v_L$ ) sound velocity are parameters of equation of motion of the three-dimensional elastic body. Transverse and longitudinal sound velocity are directly connected to stiffness constants and the mass density. If we assume that nanoparticles are small spheres, equation of motion must be solved in spherical coordinate. It is useful to introduce dimensionless variables  $\eta = \omega R/v_T = \omega d/2v_T$  and  $\xi = \omega R/v_L = \omega d/2v_L$  where  $R$  is radius of the particle ( $d$  is diameter). Boundary conditions definitely determine the solutions [14–18].

If one assumes that there is no displacement at the particle surface, it is so-called rigid boundary conditions. If one assumes that there is no force on the surface, i.e. radial components of the stress tensor at the surface are zero, it is so-called free-surface boundary conditions. We assumed the stress-free boundary conditions [17]. We believe that this model is adequate to describe behavior of nanoparticles in our samples, as there is no compact matrix surrounding them.

In spherical case, each value of angular momentum quantum number  $l$  ( $l=0,1,2,\dots$ ) gives a series of solutions. We numerate these solutions  $n$  ( $n=0,1,2,\dots$ ). So, eigen solutions are labeled as  $\eta^{nl}$  ( $\xi^{nl}$ ), and eigen states are labeled as  $(n,l)$ . Two types of vibrational modes are obtained: spheroidal and torsional modes. As  $\omega^{nl} = 2\eta^{nl}v_T/d$  i.e.  $\omega^{nl} = 2\xi^{nl}v_L/d$ , each solution gives one linear dependence  $\omega^{nl} = f(1/d)$ .

Dimensionless solutions of equation for spheroidal modes strongly depend on the material through ratio  $v_L/v_T$ . Solutions for torsional modes, do not depend on material. According to the group theory analysis the spheroidal  $l=0$  and  $l=2$  modes are Raman active [19]. Theory implies [20] that the spheroidal mode  $(0,0)$  is the most intensive in the  $l=0$  series, and the most intensive of all Raman active modes. In a lot of experimental spectra only this mode was detected. The first quadrupolar mode is the most intensive in  $l=2$  series. The frequency of  $(0,2)$  mode is almost always lower than the  $(0,0)$  mode frequency.

Structure of  $\text{ZnFe}_2\text{O}_4$  is similar to  $\text{Fe}_3\text{O}_4$ . In analysis of  $\text{ZnFe}_2\text{O}_4$  we started from  $\text{Fe}_3\text{O}_4$  acoustic properties. For  $\text{Fe}_3\text{O}_4$  of spinel (cubic) structure we used following parameters: mass



**Fig. 4.** (a) Dependence of eigen solutions, for spheroidal modes in the surface stress-free approximation, on  $\text{Fe}_2\text{O}_3$  nanoparticles inverse diameter:  $\nu_L = 6400$  m/s and  $\nu_T = 2960$  m/s. Plotted lines represent eigen solutions. Solid and opened cycles are experimental results of  $\text{ZnFe}_2\text{O}_4$  and  $\text{Fe}_2\text{O}_3$  respectively. (b) Dependence of eigen solutions on nanoparticles inverse diameter:  $\nu_L = 8355$  m/s and  $\nu_T = 4397$  m/s. Solid cycles are experimental results of  $\text{ZnFe}_2\text{O}_4$ .

density  $5240 \text{ kg/m}^3$  and stiffness constants  $C_{11} = 2.17 \times 10^{11} \text{ N/m}^2$ ,  $C_{12} = 1.21 \times 10^{11} \text{ N/m}^2$ ,  $C_{44} = 0.46 \times 10^{11} \text{ N/m}^2$  [21]. Longitudinal and transverse sound velocities in a case of cubic crystal are:  $\nu_L = (C_{11}/\rho)^{1/2} = 6400$  m/s and  $\nu_T = (C_{44}/\rho)^{1/2} = 2960$  m/s. Solutions for  $\text{Fe}_3\text{O}_4$ , in case of stress-free boundary conditions, for the  $l=0, n=0,1,2$  and  $l=2, n=0,1,2,3,4$  are presented in Fig. 4(a). We marked (solid cycles) positions of frequencies that are result of deconvolution of  $\text{ZnFe}_2\text{O}_4$  particles experimental spectra. First two frequencies,  $12.6 \text{ cm}^{-1}$  and  $24 \text{ cm}^{-1}$ , that are the best defined structures in spectra, are located very close to  $\text{Fe}_3\text{O}_4$  (0,0) and (0,2) solutions, i.e. lines in Fig. 4(a). Also mode at  $\approx 24 \text{ cm}^{-1}$  is more intensive than mode at  $\approx 12.6 \text{ cm}^{-1}$ , Fig. 3(a).

We tentatively attributed mode at  $24 \text{ cm}^{-1}$ , Fig. 3(a), to be  $\omega^{00}$  (0,0) and mode at  $12.6 \text{ cm}^{-1}$ , Fig. 3(a), to be  $\omega^{02}$  (0,2). As  $\omega^{00} = 2\xi^{00}\nu_L/d$  and  $\omega^{02} = 2\eta^{02}\nu_T/d = 2\xi^{02}\nu_L/d$  follows that  $\omega^{00}/\omega^{02} = \xi^{00}/\xi^{02} = 1.9$ . From curves  $\xi^{00}(\nu_L/\nu_T)$  and  $\xi^{02}(\nu_L/\nu_T)$  function  $g(\nu_L/\nu_T) = \xi^{00}(\nu_L/\nu_T)/\xi^{02}(\nu_L/\nu_T)$  can be established. As  $\omega^{00}/\omega^{02} = 1.9$ , from  $g(\nu_L/\nu_T)$  follows that  $\nu_L/\nu_T = 1.9$  and  $\xi^{00}(1.9) = 2.68$ . From  $\omega^{00} = 24 \text{ cm}^{-1}$   $d = 10 \text{ nm}$  follows:  $\nu_L = 8355$  m/s and  $\nu_T = \nu_L/1.9 = 4397$  m/s. These are new parameters that can be prescribed to  $\text{ZnFe}_2\text{O}_4$ . We solved complete problem with these new parameters. New solutions are as lines presented in Fig. 4(b).

Basic assumption was that  $\omega^{00} = 24 \text{ cm}^{-1}$  and  $\omega^{02} = 12.6 \text{ cm}^{-1}$ . As we see  $\omega^{12} \approx \omega^{00}$ , Fig. 4(b) and cannot be separated and detected in experimental spectra. Band at  $\approx 40 \text{ cm}^{-1}$  is identified as (2,2) and band at  $\approx 54 \text{ cm}^{-1}$  as (1,0) mode. Very good agreement between experimental and calculated results, Fig. 4(b), imply that our assumptions were correct.

In Fig. 4(a) we marked (open cycles) frequency positions that are result of experimental  $\text{Fe}_2\text{O}_3$  particles spectra deconvolution ( $d = 24 \text{ nm}$ ) Fig. 3(b). As expected experimental frequencies are far away from calculated  $\text{Fe}_3\text{O}_4$  values, Fig. 4(a). For cubic system it is reasonable to use effective i.e. averaged parameters, as we did for  $\text{Fe}_3\text{O}_4$  and  $\text{ZnFe}_2\text{O}_4$ .  $\alpha\text{-Fe}_2\text{O}_3$  crystallize in rhombohedral (trigonal)

system, conventionally viewed as hexagonal. Stiffness constants, and consequently sound velocities, differ very much for different crystallographic axes in  $\alpha\text{-Fe}_2\text{O}_3$  and from  $\text{Fe}_3\text{O}_4$  values. It was not reasonable to follow the procedure with averaged parameters as in  $\text{ZnFe}_2\text{O}_4$ . To established parameter of material that determined low-frequency Raman spectra of  $\text{Fe}_2\text{O}_3$  more detailed analysis is needed.

#### 4. Conclusion

Small amount (5 wt.%) of  $\text{Fe}_2\text{O}_3$  at the beginning of the synthesis results in forming of  $\text{ZnFe}_2\text{O}_4$  nanoparticles. Large amount (90 wt.%) of  $\text{Fe}_2\text{O}_3$  at the beginning of the synthesis results in forming  $\text{Fe}_2\text{O}_3$  nanoparticles. Both samples contain  $\text{ZnO}$  phase which is not registered by XRD, but is clearly seen in the Raman spectra. In low-frequency Raman spectra of  $\text{ZnFe}_2\text{O}_4$  nanoparticles registered peaks agree well with the calculated frequencies of acoustic phonons. As a result we identified (0,2), (0,0), (2,2) and (1,0) modes

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#### References

- [1] Y. Chen, D.M. Bagnall, H. Koh, K. Park, K. Higara, Z. Zhu, T. Yao, J. Appl. Phys. 84 (1988) 3912.
- [2] J. Nemeth, G. Rodriguez-Gattorno, A. Diaz, I. Dekany, Langmuir 20 (2004) 2855.
- [3] J.M.D. Coey, M. Venkatesan, C.B. Fitzgerald, Nat. Mater. 4 (2005) 173.

- [4] C. Sudakar, J.S. Thakur, G. Lawes, R. Naik, V.M. Naik, *Phys. Rev. B* 75 (2007) 054423.
- [5] T. Dietl, *Acta Phys. Pol. A* 111 (2007) 27.
- [6] U. Narkiewicz, D. Sibera, I. Kuryliszyn-Kudelska, L. Kilanski, W.N. Dobrowolski, *Romcevic Acta Phys. Pol. A* 113 (2008) 1695.
- [7] A.L. Patterson, *Phys. Rev.* 56 (1993) 978.
- [8] N. Ashkenov, B.N. Mbenkum, C. Bundesmann, V. Riede, M. Lorenz, D. Spemann, E.M. Kaidashev, A. Kasic, M. Shubert, M. Grundmann, *J. Appl. Phys.* 93 (2003) 126.
- [9] E.F. Venger, A.V. Melnichuk, L. Lu Melnichuk, Yu A. Pasechuk, *Phys. Status Solidi B* 188 (1995) 823.
- [10] Z. Wang, D. Schiferl, Y. Zhao, H.St.C. O'Neill, *J. Phys. Chem. Solids* 64 (2003) 2517.
- [11] F.Bauciuman, F. Patcas, R. Craciun, D.R.T. Zahn, *Phys. Chem. Chem. Phys.* 1 (1999) 185.
- [12] O.N. Shebanova, P. Lazar, J. Solid State Chem. 174 (2003) 424.
- [13] V.G. Hajdiev, M.N. Iliev, I.V. Vergilov, *J. Phys. C: Solid State Phys.* 21 (1988) 1199.
- [14] A. Tanaka, S. Onari, T. Arai, *Phys. Rev. B* 47 (1993) 1237.
- [15] L. Saviot, B. Champagnon, E. Duval, A.I. Ekimov, *Phys. Rev. B* 57 (1998) 341.
- [16] L. Saviot, D.B. Murray, M. del, C. Marco de Lucas, *Phys. Rev. B* 69 (2004) 113402.
- [17] L. Saviot, B. Champagnon, E. Duval, I.A. Kudriavtsev, I.E. Akimov, *J. Non-Cryst. Solids* 197 (1996) 238.
- [18] E. Roca, C. Trallero-Giner, M. Cardona, *Phys. Rev. B* 49 (1994) 13704.
- [19] E. Duval, *Phys. Rev. B* 46 (1992) 5795.
- [20] N. Combe, J.R. Huntzinger, A. Mlayah, *Phys. Rev. B* 76 (2007) 205425.
- [21] P. Nordlander, M. Ronay, *Phys. Rev. B* 36 (1987) 4982.